

| Course syllabus for First cycle studies | | | | | | | | | |
|---|--|--|------------------------------------|-----------------------|----------|--|--|--|--|
| 1. | Course title | General and Inorganic Chemistry 1 | | | | | | | |
| 2. | Code | MDE1M2 | | | | | | | |
| 3. | Study Program | Metallurgical Digital Engineering | | | | | | | |
| 4. | Study program organizer (unit, institute, department, division) | | | | | | | | |
| 5. | Degree (first, second, third cycle) | First degree | | | | | | | |
| 6. | Academic year / semester | I semester | 7. | Number of ECTS | 7 | | | | |
| 8. | Instructors | Gordana Ruseska, full professor Biljana Angjuseva, full professor | | | | | | | |
| 9. | Prerequisites for course enrollment | / | | | | | | | |
| 10. | Objectives of the course syllabus (competences): Acquired skills (competences): | | | | | | | | |
| 11. | Content of the course: Chemistry as a Natural Science Types of Matter Laws for chemical combination of elements by mass Law for chemical combination of elements by volume Gas laws Modern concept of chemical element Atomic and molecular mass Electronic configuration Types of chemical bonds (ionic, covalent and metallic bonds) Thermochemical equations (heat effects of chemical processes) Rate of chemical reaction | | | | | | | | |
| 12. | Study methods: | | | | | | | | |
| 13. | Total available time | | | | | | | | |
| 14. | Allocation of available time | | | | | | | | |
| 15. | Teaching activities | 15.1. | Lectures | | 45 hours | | | | |
| | | 15.2. | Labotatory and Classroom Exercises | | 45 hours | | | | |
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| 16. | Other types of activities | 16.1. | Project Tasks | | 20 hours | | | | |
| | | 16.2. | Independent Tasks | | 40 hours | | | | |
| | | 16.3. | Homework | | 90 hours | | | | |
| 17. | Grading system | | | | | | | | |
| | 17.1. | | | | | | | | |
| | 17.2. | | | | | | | | |
| | 17.3. | | | | | | | | |
| 18. | Grading criteria | Up to 61 points | | 5 (five) (F) | | | | | |

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|-----|--|-----------------------|---|--|------------------------------|------|
| | (points/grade) | From 61 to 69 points | 6 (six) (E) | | | |
| | | From 70 to 79 points | 7 (seven) (D) | | | |
| | | from 80 to 89 points | 8 (eight) (C) | | | |
| | | From 90 to 95 points | 9 (nine) (B) | | | |
| | | from 95 to 100 points | 10 (ten) (A) | | | |
| 19. | Prerequisites for taking the final exam | | | | | |
| 20. | Language in which lectures are conducted | | | | | |
| 21. | Method for monitoring the quality of lectures | | | | | |
| 22. | LITERATURE | | | | | |
| | 22.1. | Compulsory literature | | | | |
| | | No. | Author | Title | Publisher | Year |
| | | 1. | N.N. Greenwood A. Earnshaw | Chemistry of the elements (Second Edition) ISBN: 978075063 3659 | | 1997 |
| | | 2. | Raymond Chang, Jason Overby ISBN:978007 6812141 | CHEMISTRY Edition 13 | MC graw Hill Education | 2018 |
| | 22.2. | Additional literature | | | | |
| | | No. | Author | Title | Publisher | Year |
| | | 1. | A. Smith | Intro- duction to general inorganic chemistry | 2012 | |
| | | 2. | | | | |
| | | 3. | | | | |